

molecules whose water solutions pdf

The molecule of water. A molecule is an aggregation of atomic nuclei and electrons that is sufficiently stable to possess observable properties and there are few molecules that are more stable and difficult to decompose than H₂O.

Water and its structure Chem1

BUFFER SOLUTIONS - INTRODUCTION AND USES Definition Solutions which resist changes in pH when small quantities of acid or alkali are added.
• Types Acidic Buffer (pH < 7) weak acid + its sodium or potassium salt

BUFFER SOLUTIONS - INTRODUCTION AND USES

Intermolecular forces (IMF) are the forces which mediate interaction between molecules, including forces of attraction or repulsion which act between molecules and other types of neighboring particles, e.g., atoms or ions.

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